

# Periodensystem der Elemente

|                                                                                            |  |                                                                                               |  |                                                                                                        |  |                                                                                                                    |  |                                                                                                                        |  |                                                                                                                  |  |                                                                                                               |  |                                                                                                                  |  |                                                                                                               |  |                                                                                                              |  |                                                                                                             |  |                                                                                                                  |  |                                                                                                                                |  |                                                                                                                            |  |                                                                                                                            |  |                                                                                                                             |  |                                                                                                                        |  |                                                                                                                        |  |          |  |         |  |        |  |         |  |          |  |
|--------------------------------------------------------------------------------------------|--|-----------------------------------------------------------------------------------------------|--|--------------------------------------------------------------------------------------------------------|--|--------------------------------------------------------------------------------------------------------------------|--|------------------------------------------------------------------------------------------------------------------------|--|------------------------------------------------------------------------------------------------------------------|--|---------------------------------------------------------------------------------------------------------------|--|------------------------------------------------------------------------------------------------------------------|--|---------------------------------------------------------------------------------------------------------------|--|--------------------------------------------------------------------------------------------------------------|--|-------------------------------------------------------------------------------------------------------------|--|------------------------------------------------------------------------------------------------------------------|--|--------------------------------------------------------------------------------------------------------------------------------|--|----------------------------------------------------------------------------------------------------------------------------|--|----------------------------------------------------------------------------------------------------------------------------|--|-----------------------------------------------------------------------------------------------------------------------------|--|------------------------------------------------------------------------------------------------------------------------|--|------------------------------------------------------------------------------------------------------------------------|--|----------|--|---------|--|--------|--|---------|--|----------|--|
| 1 / I                                                                                      |  |                                                                                               |  |                                                                                                        |  |                                                                                                                    |  |                                                                                                                        |  |                                                                                                                  |  |                                                                                                               |  |                                                                                                                  |  |                                                                                                               |  | 18 / IIX                                                                                                     |  |                                                                                                             |  |                                                                                                                  |  |                                                                                                                                |  |                                                                                                                            |  |                                                                                                                            |  |                                                                                                                             |  |                                                                                                                        |  |                                                                                                                        |  |          |  |         |  |        |  |         |  |          |  |
| 1<br><b>H</b><br>1s <sup>1</sup><br>Wasserstoff<br>-259 2.2<br>-253 0.08<br>-1,1           |  |                                                                                               |  |                                                                                                        |  |                                                                                                                    |  |                                                                                                                        |  |                                                                                                                  |  |                                                                                                               |  |                                                                                                                  |  |                                                                                                               |  | 2<br><b>He</b><br>1s <sup>2</sup><br>Helium<br>-272 -<br>-269 0.17                                           |  |                                                                                                             |  |                                                                                                                  |  |                                                                                                                                |  |                                                                                                                            |  |                                                                                                                            |  |                                                                                                                             |  |                                                                                                                        |  |                                                                                                                        |  |          |  |         |  |        |  |         |  |          |  |
| 2 / II                                                                                     |  | 3                                                                                             |  |                                                                                                        |  |                                                                                                                    |  |                                                                                                                        |  |                                                                                                                  |  |                                                                                                               |  |                                                                                                                  |  |                                                                                                               |  | 4                                                                                                            |  | 5                                                                                                           |  | 6                                                                                                                |  | 7                                                                                                                              |  | 8                                                                                                                          |  | 9                                                                                                                          |  | 10                                                                                                                          |  | 11                                                                                                                     |  | 12                                                                                                                     |  | 13 / III |  | 14 / IV |  | 15 / V |  | 16 / VI |  | 17 / VII |  |
| 3<br><b>Li</b><br>1s <sup>2</sup> 2s <sup>1</sup><br>Lithium<br>181 0.97<br>1342 0.53<br>1 |  | 4<br><b>Be</b><br>1s <sup>2</sup> 2s <sup>2</sup><br>Beryllium<br>1287 1.47<br>2469 1.85<br>2 |  | 5<br><b>B</b><br>1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>1</sup><br>Bor<br>2075 2.01<br>3927 2.46<br>3 |  | 6<br><b>C</b><br>1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup><br>Kohlenstoff<br>3550 2.5<br>4827 3.51<br>-4,2,4 |  | 7<br><b>N</b><br>1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup><br>Stickstoff<br>-210 3.07<br>-196 1.17<br>-3,2,3,4,5 |  | 8<br><b>O</b><br>1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup><br>Sauerstoff<br>-219 3.5<br>-183 1.33<br>-2,-1 |  | 9<br><b>F</b><br>1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>5</sup><br>Fluor<br>-220 4.1<br>-188 1.58<br>-1      |  | 10<br><b>Ne</b><br>1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup><br>Neon<br>-249 -<br>-246 0.84                |  | 11<br><b>Na</b><br>[Ne]3s <sup>1</sup><br>Natrium<br>98 1.01<br>883 0.97<br>1                                 |  | 12<br><b>Mg</b><br>[Ne]3s <sup>2</sup><br>Magnesium<br>650 1.23<br>1090 1.74<br>2                            |  | 13<br><b>Al</b><br>[Ne]3s <sup>2</sup> 3p <sup>1</sup><br>Aluminium<br>660 1.47<br>2519 2.7<br>3            |  | 14<br><b>Si</b><br>[Ne]3s <sup>2</sup> 3p <sup>2</sup><br>Silizium<br>1414 1.74<br>2365 2.33<br>-4,4             |  | 15<br><b>P</b><br>[Ne]3s <sup>2</sup> 3p <sup>3</sup><br>Phosphor<br>44 2.06<br>277 1.82<br>-3,3,5                             |  | 16<br><b>S</b><br>[Ne]3s <sup>2</sup> 3p <sup>4</sup><br>Schwefel<br>115 2.44<br>445 2.06<br>-2,2,4,6                      |  | 17<br><b>Cl</b><br>[Ne]3s <sup>2</sup> 3p <sup>5</sup><br>Chlor<br>-102 2.83<br>-34 2.95<br>-1,1,3,5,7                     |  | 18<br><b>Ar</b><br>[Ne]3s <sup>2</sup> 3p <sup>6</sup><br>Argon<br>-189 -<br>-186 1.66                                      |  |                                                                                                                        |  |                                                                                                                        |  |          |  |         |  |        |  |         |  |          |  |
| 19<br><b>K</b><br>[Ar]4s <sup>1</sup><br>Kalium<br>63 0.91<br>759 0.86<br>1                |  | 20<br><b>Ca</b><br>[Ar]4s <sup>2</sup><br>Calcium<br>842 1.04<br>1484 1.54<br>2               |  | 21<br><b>Sc</b><br>[Ar]3d <sup>1</sup> 4s <sup>2</sup><br>Scandium<br>1541 1.2<br>2836 2.99<br>3       |  | 22<br><b>Ti</b><br>[Ar]3d <sup>2</sup> 4s <sup>2</sup><br>Titan<br>1668 1.32<br>3287 4.51<br>3,4                   |  | 23<br><b>V</b><br>[Ar]3d <sup>3</sup> 4s <sup>2</sup><br>Vanadium<br>1910 1.45<br>3407 6.09<br>0,2,3,6                 |  | 24<br><b>Cr</b><br>[Ar]3d <sup>5</sup> 4s <sup>1</sup><br>Chrom<br>1907 1.56<br>2671 7.14<br>0,2,3,6             |  | 25<br><b>Mn</b><br>[Ar]3d <sup>5</sup> 4s <sup>2</sup><br>Mangan<br>1246 1.6<br>2061 7.44<br>-1,0,2,3,4,6,7   |  | 26<br><b>Fe</b><br>[Ar]3d <sup>6</sup> 4s <sup>2</sup><br>Eisen<br>1538 1.64<br>2861 7.87<br>-2,0,2,3,6          |  | 27<br><b>Co</b><br>[Ar]3d <sup>7</sup> 4s <sup>2</sup><br>Cobalt<br>1495 1.7<br>2927 8.89<br>-1,0,2,3         |  | 28<br><b>Ni</b><br>[Ar]3d <sup>8</sup> 4s <sup>2</sup><br>Nickel<br>1455 1.75<br>2913 8.91<br>0,2,3          |  | 29<br><b>Cu</b><br>[Ar]3d <sup>10</sup> 4s <sup>1</sup><br>Kupfer<br>1085 1.75<br>2562 8.92<br>1,2          |  | 30<br><b>Zn</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup><br>Zink<br>420 1.66<br>907 7.14<br>2                     |  | 31<br><b>Ga</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>1</sup><br>Gallium<br>30 1.82<br>2204 5.91<br>3                |  | 32<br><b>Ge</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup><br>Germanium<br>938 2.02<br>2833 5.32<br>4         |  | 33<br><b>As</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup><br>Arsen<br>615 2.2<br>615 5.72<br>-3,3,5          |  | 34<br><b>Se</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup><br>Selen<br>221 2.48<br>685 4.82<br>-2,4,6          |  | 35<br><b>Br</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup><br>Brom<br>-7 2.74<br>59 3.14<br>-1,1,3,5,7    |  | 36<br><b>Kr</b><br>[Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup><br>Krypton<br>-157 -<br>-153 3.48<br>2,4       |  |          |  |         |  |        |  |         |  |          |  |
| 37<br><b>Rb</b><br>[Kr]5s <sup>1</sup><br>Rubidium<br>39 0.89<br>688 1.53<br>1             |  | 38<br><b>Sr</b><br>[Kr]5s <sup>2</sup><br>Strontium<br>777 0.99<br>1382 2.63<br>2             |  | 39<br><b>Y</b><br>[Kr]4d <sup>1</sup> 5s <sup>2</sup><br>Yttrium<br>1526 1.11<br>3336 4.47<br>3        |  | 40<br><b>Zr</b><br>[Kr]4d <sup>2</sup> 5s <sup>2</sup><br>Zirkonium<br>1855 1.22<br>4409 6.51<br>4                 |  | 41<br><b>Nb</b><br>[Kr]4d <sup>4</sup> 5s <sup>1</sup><br>Niob<br>2477 1.23<br>4744 8.58<br>3,5                        |  | 42<br><b>Mo</b><br>[Kr]4d <sup>5</sup> 5s <sup>1</sup><br>Molybdän<br>2623 1.3<br>4639 10.28<br>0,2,3,4,5,6      |  | 43<br><b>Tc</b><br>[Kr]4d <sup>5</sup> 5s <sup>2</sup><br>Technetium<br>2157 1.36<br>4265 11.49<br>7          |  | 44<br><b>Ru</b><br>[Kr]4d <sup>7</sup> 5s <sup>1</sup><br>Ruthenium<br>2334 1.42<br>4150 12.45<br>-2,0,2,3,4,6,8 |  | 45<br><b>Rh</b><br>[Kr]4d <sup>8</sup> 5s <sup>1</sup><br>Rhodium<br>1964 1.45<br>3695 12.41<br>0,1,2,3,4,5   |  | 46<br><b>Pd</b><br>[Kr]4d <sup>10</sup><br>Palladium<br>1555 1.3<br>2963 12.02<br>0,2,4                      |  | 47<br><b>Ag</b><br>[Kr]4d <sup>10</sup> 5s <sup>1</sup><br>Silber<br>962 1.42<br>2162 10.49<br>1,2          |  | 48<br><b>Cd</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup><br>Cadmium<br>321 1.46<br>767 8.64<br>2                  |  | 49<br><b>In</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>1</sup><br>Indium<br>157 1.49<br>2072 7.31<br>3                |  | 50<br><b>Sn</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup><br>Zinn<br>232 1.72<br>2602 7.29<br>2,4            |  | 51<br><b>Sb</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>3</sup><br>Antimon<br>631 1.82<br>1587 6.69<br>-3,3,5      |  | 52<br><b>Te</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>4</sup><br>Tellur<br>450 2.01<br>988 6.25<br>-2,4,6         |  | 53<br><b>I</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>5</sup><br>Iod<br>114 2.21<br>184 4.94<br>-1,1,3,5,7    |  | 54<br><b>Xe</b><br>[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>6</sup><br>Xenon<br>-112 -<br>-108 4.49<br>2,4,6       |  |          |  |         |  |        |  |         |  |          |  |
| 55<br><b>Cs</b><br>[Xe]6s <sup>1</sup><br>Cäsium<br>28 0.86<br>671 1.09<br>1               |  | 56<br><b>Ba</b><br>[Xe]6s <sup>2</sup><br>Barium<br>727 0.97<br>1897 3.65<br>2                |  | 57<br><b>La</b><br>[Xe]5d <sup>1</sup> 6s <sup>2</sup><br>Lanthan<br>920 1.08<br>3464 6.16<br>3        |  | 58<br><b>Hf</b><br>[Xe]4f <sup>14</sup> 5d <sup>2</sup> 6s <sup>2</sup><br>Hafnium<br>2233 1.23<br>4603 13.31      |  | 59<br><b>Ta</b><br>[Xe]4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup><br>Tantal<br>3017 1.33<br>5458 16.68           |  | 60<br><b>W</b><br>[Xe]4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup><br>Wolfram<br>3422 1.4<br>5555 19.26      |  | 61<br><b>Re</b><br>[Xe]4f <sup>14</sup> 5d <sup>5</sup> 6s <sup>2</sup><br>Rhenium<br>3186 1.46<br>5596 21.03 |  | 62<br><b>Os</b><br>[Xe]4f <sup>14</sup> 5d <sup>6</sup> 6s <sup>2</sup><br>Osmium<br>3033 1.52<br>5012 22.61     |  | 63<br><b>Ir</b><br>[Xe]4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup><br>Iridium<br>2446 1.55<br>4428 22.65 |  | 64<br><b>Pt</b><br>[Xe]4f <sup>14</sup> 5d <sup>9</sup> 6s <sup>1</sup><br>Platin<br>1768 1.42<br>3825 21.45 |  | 65<br><b>Au</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>1</sup><br>Gold<br>1064 1.42<br>2856 19.32 |  | 66<br><b>Hg</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup><br>Quecksilber<br>-39 1.44<br>357 13.55 |  | 67<br><b>Tl</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>1</sup><br>Thallium<br>304 1.44<br>1473 11.85 |  | 68<br><b>Pb</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>2</sup><br>Blei<br>327 1.55<br>1749 11.34 |  | 69<br><b>Bi</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>3</sup><br>Bismut<br>272 1.67<br>1564 9.8 |  | 70<br><b>Po</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>4</sup><br>Polonium<br>254 1.76<br>962 9.2 |  | 71<br><b>At</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>5</sup><br>Astat<br>302 1.96<br>337 - |  | 72<br><b>Rn</b><br>[Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>6</sup><br>Radon<br>-71 -<br>-62 9.23 |  |          |  |         |  |        |  |         |  |          |  |
| 87<br><b>Fr</b><br>[Rn]7s <sup>1</sup><br>Francium<br>27 0.86<br>677 -                     |  | 88<br><b>Ra</b><br>[Rn]7s <sup>2</sup><br>Radium<br>700 0.97<br>1737 5.5                      |  | 89<br><b>Ac</b><br>[Rn]6d <sup>1</sup> 7s <sup>2</sup><br>Actinium<br>1050 1<br>3198 10.07             |  | 90<br><b>Rf</b><br>[Rn]5f <sup>14</sup> 6d <sup>2</sup> 7s <sup>2</sup><br>Rutherfordium<br>- - -                  |  | 91<br><b>Db</b><br>[Rn]5f <sup>14</sup> 6d <sup>3</sup> 7s <sup>2</sup><br>Dubnium<br>- - -                            |  | 92<br><b>Sg</b><br>[Rn]5f <sup>14</sup> 6d <sup>4</sup> 7s <sup>2</sup><br>Seaborgium<br>- - -                   |  | 93<br><b>Bh</b><br>[Rn]5f <sup>14</sup> 6d <sup>5</sup> 7s <sup>2</sup><br>Bohrium<br>- - -                   |  | 94<br><b>Hs</b><br>[Rn]5f <sup>14</sup> 6d <sup>6</sup> 7s <sup>2</sup><br>Hassium<br>- - -                      |  | 95<br><b>Mt</b><br>[Rn]5f <sup>14</sup> 6d <sup>7</sup> 7s <sup>2</sup><br>Meitnerium<br>- - -                |  | 96<br><b>Ds</b><br>[Rn]5f <sup>14</sup> 6d <sup>8</sup> 7s <sup>2</sup><br>Darmstadtium<br>- - -             |  | 97<br><b>Rg</b><br>[Rn]5f <sup>14</sup> 6d <sup>9</sup> 7s <sup>2</sup><br>Roentgenium<br>- - -             |  | 98<br><b>Cn</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup><br>Copernicium<br>- - -                 |  | 99<br><b>Nh</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 6p <sup>1</sup><br>Nihonium<br>- - -                  |  | 100<br><b>Fl</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 6p <sup>2</sup><br>Fleivorium<br>- - -           |  | 101<br><b>Mc</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 6p <sup>3</sup><br>Moscovium<br>- - -            |  | 102<br><b>Lv</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 6p <sup>4</sup><br>Livermorium<br>- - -           |  | 103<br><b>Ts</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 6p <sup>5</sup><br>Tennessin<br>- - -        |  | 104<br><b>Og</b><br>[Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 6p <sup>6</sup><br>Oganesson<br>- - -        |  |          |  |         |  |        |  |         |  |          |  |

### \*Aggregatzustand:

Die Farbe des Elementsymbols signalisiert

gleichzeitig den Aggregatzustand bei

Raumtemperatur (20 °C)

Schwarz: fest

Blau: flüssig

Rot: gasförmig

Grau: keine Angabe

### \*\*Dichte:

Für die meisten Elemente ist die Dichte in g/cm<sup>3</sup>

angegeben.

Für Gase (Rotes Elementsymbol) ist die Dichte in g/l

angegeben.

|                                                                                                                      |                                                                                                                          |                                                                                                                 |                                                                                                                      |                                                                                                            |                                                                                                           |                                                                                                                            |                                                                                                   |                                                                                                     |                                                                                                    |                                                                                                    |                                                                                                     |                                                                                                      |                                                                                                                      |
|----------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------------------------|
| 58 140.12<br><b>Ce</b><br>[Xe]4f <sup>5</sup> 5d <sup>1</sup> 6s <sup>2</sup><br>Cer<br>795 1.08<br>3443 6.77<br>3,4 | 59 140.91<br><b>Pr</b><br>[Xe]4f <sup>6</sup> 6s <sup>2</sup><br>Praseodym<br>935 1.07<br>3520 6.48<br>3,4               | 60 144.24<br><b>Nd</b><br>[Xe]4f <sup>6</sup> 6s <sup>2</sup><br>Neodym<br>1024 1.07<br>3074 7<br>3             | 61 144.91<br><b>Pm</b><br>[Xe]4f <sup>6</sup> 6s <sup>2</sup><br>Promethium<br>1042 1.07<br>3000 7.22<br>3           | 62 150.36<br><b>Sm</b><br>[Xe]4f <sup>6</sup> 6s <sup>2</sup><br>Samarium<br>1072 1.07<br>1794 7.54<br>2,3 | 63 151.96<br><b>Eu</b><br>[Xe]4f <sup>7</sup> 6s <sup>2</sup><br>Europium<br>826 1.01<br>1529 5.25<br>2,3 | 64 157.25<br><b>Gd</b><br>[Xe]4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup><br>Gadolinium<br>1312 1.11<br>3273 7.89<br>3 | 65 158.93<br><b>Tb</b><br>[Xe]4f <sup>9</sup> 6s <sup>2</sup><br>Terbium<br>1356 1.1<br>3230 8.25 | 66 162.5<br><b>Dy</b><br>[Xe]4f <sup>9</sup> 6s <sup>2</sup><br>Dysprosium<br>1407 1.1<br>2567 8.56 | 67 164.93<br><b>Ho</b><br>[Xe]4f <sup>10</sup> 6s <sup>2</sup><br>Holmium<br>1461 1.1<br>2720 8.78 | 68 167.26<br><b>Er</b><br>[Xe]4f <sup>10</sup> 6s <sup>2</sup><br>Erbium<br>1529 1.11<br>2868 9.05 | 69 168.93<br><b>Tm</b><br>[Xe]4f <sup>11</sup> 6s <sup>2</sup><br>Thulium<br>1545 1.11<br>1950 9.32 | 70 173.05<br><b>Yb</b><br>[Xe]4f <sup>12</sup> 6s <sup>2</sup><br>Ytterbium<br>824 1.06<br>1196 6.97 | 71 174.97<br><b>Lu</b><br>[Xe]4f <sup>13</sup> 5d <sup>1</sup> 6s <sup>2</sup><br>Lutetium<br>1652 1.14<br>3402 9.84 |
| 90 232.04<br><b>Th</b><br>[Rn]6d <sup>2</sup> 7s <sup>2</sup><br>Thorium<br>1842 1.11<br>4788 11.72                  | 91 231.04<br><b>Pa</b><br>[Rn]5f <sup>6</sup> 6d <sup>1</sup> 7s <sup>2</sup><br>Protactinium<br>1568 1.14<br>4027 15.37 | 92 238.03<br><b>U</b><br>[Rn]5f <sup>6</sup> 6d <sup>1</sup> 7s <sup>2</sup><br>Uran<br>1132 1.22<br>4131 18.97 | 93 237.05<br><b>Np</b><br>[Rn]5f <sup>6</sup> 6d <sup>1</sup> 7s <sup>2</sup><br>Neptunium<br>644 1.22<br>4000 20.48 | 94 244.06<br><b>Pu</b><br>[Rn]5f <sup>7</sup> 7s <sup>2</sup><br>Plutonium<br>639 1.22<br>3228 19.74       | 95 243.06<br><b>Am</b><br>[Rn]5f <sup>7</sup> 7s <sup>2</sup><br>Americium<br>1176 1.2<br>2607 13.67      | 96 247.07<br><b>Cm</b><br>[Rn]5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup><br>Curium<br>1340 1.2<br>3110 13.51          | 97 247.07<br><b>Bk</b><br>[Rn]5f <sup>7</sup> 7s <sup>2</sup><br>Berkelium<br>1050 1.2<br>- 13.25 | 98 251.08<br><b>Cf</b><br>[Rn]5f <sup>9</sup> 7s <sup>2</sup><br>Californium<br>900 1.2<br>- 15.1   | 99 252.08<br><b>Es</b><br>[Rn]5f <sup>9</sup> 7s <sup>2</sup><br>Einsteinium<br>860 1.2<br>- -     | 100 257.1<br><b>Fm</b><br>[Rn]5f <sup>9</sup> 7s <sup>2</sup><br>Fermium<br>1527 1.2<br>- -        | 101 258.1<br><b>Md</b><br>[Rn]5f <sup>9</sup> 7s <sup>2</sup><br>Mendelevium<br>827 1.2<br>- -      | 102 259.1<br><b>No</b><br>[Rn]5f <sup>9</sup> 7s <sup>2</sup><br>Nobelium<br>827 1.2<br>- -          | 103 262.11<br><b>Lr</b><br>[Rn]5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup><br>Lawrencium<br>1627 -<br>- -       |